



End-Permian extinction amplified by plume-induced release of recycled lithospheric volatiles

DOI:

[10.1038/s41561-018-0215-4](https://doi.org/10.1038/s41561-018-0215-4)

Document Version

Accepted author manuscript

[Link to publication record in Manchester Research Explorer](#)

Citation for published version (APA):

Broadley, M., Barry, P. H., Ballentine, C. J., Taylor, L. A., & Burgess, R. (2018). End-Permian extinction amplified by plume-induced release of recycled lithospheric volatiles. *Nature Geoscience*, 11(9), 682-687. <https://doi.org/10.1038/s41561-018-0215-4>

Published in:

Nature Geoscience

Citing this paper

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15 **Magmatic volatile release to the atmosphere can lead to climatic changes and**
16 **significant environmental degradation including the production of acid rain,**
17 **ocean acidification and ozone depletion, potentially resulting in collapse of the**
18 **biosphere. The largest recorded mass extinction in Earth's history occurred at**
19 **the end-Permian, coinciding with the emplacement of the Siberian large**
20 **igneous province, suggesting that large-scale magmatism is a key driver of**
21 **global environmental change. However, the source and nature of volatiles in**
22 **the Siberian large igneous province remain contentious. Here we present**
23 **halogen compositions of sub-continental lithospheric mantle xenoliths**
24 **emplaced prior to, and after the eruption of the Siberian flood basalts. We**
25 **show that the Siberian lithosphere is massively enriched in halogens from the**
26 **infiltration of subducted seawater-derived volatiles and that a significant**
27 **amount (up to 70%) of lithospheric halogens are assimilated into the plume**
28 **and released to the atmosphere during emplacement. Plume-lithosphere**
29 **interaction is therefore a key process controlling the volatile content of large**
30 **igneous provinces and as such the extent of environmental crises, leading to**
31 **mass extinctions during their emplacement.**

32

33 Large igneous provinces (LIP) are the product of rapid eruptions of large volumes of
34 magma over short geological time scales. The Permo-Triassic Siberian flood basalts
35 (SFB) erupted $\sim 4 \times 10^6$ km³ of basalt in less than 1 Ma¹. The eruption of the SFB is
36 contemporaneous with the main stage of the end-Permian crisis and is hypothesized
37 to have contributed to environmental changes that resulted in loss of >90% of all
38 marine, and >70% of all terrestrial species^{2,3}. The end-Permian mass extinction has
39 been attributed to sharp fluctuations in global temperatures and/or increased levels

40 of ultraviolet radiation resulting from extensive ozone depletion both of which are
41 associated with the magmatic release of volatiles to the atmosphere^{2,4,7}. Yet the
42 amount of volatiles expected to be released from the SFB assuming conventional
43 plume source magmatism is insufficient to account for the environmental degradation
44 and climatic fluctuations that occurred during the end-Permian crisis, requiring an
45 additional source of volatiles to be released during SFB emplacement^{8,9}. To reconcile
46 the missing SFB volatiles, it has been variously argued for significant quantities of
47 volatiles released via contact metamorphism of a sedimentary sequence^{1,5}; melting
48 of recycled eclogite within the mantle plume¹⁰; or melting of the cratonic lithosphere⁷.
49 However, the source of volatile species responsible for climatic fluctuations and
50 ozone depletion during the end-Permian crisis remains unknown.

51

52 Here we report the first detailed halogen (Cl, Br and I) data for peridotite xenoliths
53 from two Siberian kimberlites: one (Udachnaya, 360 Ma) emplaced before, and the
54 other (Obnazhennaya, 160 Ma) after the SFB eruption ~250 Ma^{11,12} (Supplementary
55 Fig. 1). The Udachnaya peridotite xenoliths (n=9) represent melt extraction from the
56 depleted cratonic mantle; whereas Obnazhennaya xenoliths (n=6) contain both
57 Archean cratonic lithosphere and melt residues generated from the SFB plume
58 (Supplementary Information)¹³. Determining the halogen composition of the cratonic
59 sub-continental lithospheric mantle (SCLM; Udachnaya) and the SFB plume
60 residues (Obnazhennaya) provides a means to estimate the composition of the SFB
61 prior to eruption, and quantify the contribution of the lithospheric mantle to the
62 halogen budget of the SFB.

63

64 **Lithospheric mantle as a reservoir for halogens**

65 Due to its isolation and non-convective nature, the SCLM retains geochemical
66 heterogeneities introduced through interactions with mantle, crustal and subduction-
67 related sources^{14,15}. Metasomatic components infiltrating the SCLM are sampled by
68 mantle xenoliths. Rapidly transported to the surface during kimberlite volcanism,
69 these xenoliths provide a window into the composition and origin of SCLM volatiles¹⁶⁻
70 ¹⁸.

71

72 The halogen and noble gas composition of the Udachnaya and Obnazhennaya
73 xenoliths was determined using neutron-irradiated noble gas mass spectrometry
74 (Methods). The results are summarised in Supplementary Tables 2 and 3 and
75 displayed in Fig. 1. The range of Cl, Br and I concentrations within the Udachnaya
76 and Obnazhennaya xenoliths are significantly different, with the average
77 concentrations from crushing and step heating being consistently higher in
78 Udachnaya samples indicating they originate from distinct domains within the SCLM
79 (Supplementary Information). Halogen-bearing fluids present in the samples, as
80 indicated by release during crushing experiments, have a similar range of Br/Cl and
81 I/Cl in both Udachnaya and Obnazhennaya xenoliths (Fig. 2a). During stepped-
82 heating of crushed residues the xenoliths show evidence for distinct endmember
83 halogen compositions (Fig. 2b). Udachnaya retains similar Br/Cl and I/Cl as
84 measured during crushing, whilst Obnazhennaya samples have more mantle-like
85 Br/Cl and I/Cl values (Table S3).

86

87 Previously published helium isotopic data also varies between the xenoliths suites,
88 with the average $^3\text{He}/^4\text{He}$ of Udachnaya ($0.4 \pm 0.3 R_A$) consistently lower than
89 Obnazhennaya ($4.2 \pm 0.9 R_A$), which has a maximum $^3\text{He}/^4\text{He}$ ($8.4 \pm 0.3 R_A$) similar
90 to mid ocean ridge basalts (MORB)¹⁸. $^3\text{He}/^4\text{He}$, Br/Cl and I/Cl values appear to be
91 coupled (Fig. 3), showing that fluids within the Obnazhennaya xenoliths represent a
92 mixture between a component rich in radiogenic ^4He , Br and I, and a component with
93 mantle-like $^3\text{He}/^4\text{He}$ and halogen compositions. The lower $^3\text{He}/^4\text{He}$ and elevated
94 Br/Cl and I/Cl as characterised by Udachnaya xenoliths, is considered representative
95 of the ancient metasomatised section of the SCLM (metasomes) that was present
96 before significant influence of the SFB mantle plume. In contrast, Obnazhennaya
97 also contains variable amounts of a mantle-like halogen and He component. Given
98 the similarity in rare earth elements (REE) patterns and extraction age for the
99 Obnazhennaya xenoliths with regards to the SFB¹³, it is considered that mantle-like
100 volatiles were added by the Siberian plume which interacted with the lithosphere
101 having a volatile composition characterised by fluids trapped in the Udachnaya
102 xenoliths (Fig. 3).

103

104 The initial inventory of halogens within the metasomatised section of Siberian SCLM
105 prior to plume impingement can be estimated using the composition of Udachnaya
106 xenoliths. The Siberian SCLM transitions from depleted harzburgite and lherzolites to
107 predominantly metasomatised peridotites at 180-190 km¹⁹. Assuming that the
108 Udachnaya peridotite xenoliths are representative of metasomatised peridotites in
109 the lower 30 km of the SCLM, and taking the surficial area of the Siberian Craton
110 ($4 \times 10^6 \text{ km}^2$), then the metasomatised portion of the Siberian SCLM contains
111 approximately $0.6\text{-}1.5 \times 10^{19}$, $1.6\text{-}2.7 \times 10^{17}$ and $0.5\text{-}1.1 \times 10^{14}$ kg of Cl, Br and I

112 respectively (Supplementary Table 1 and Supplementary Information). The
113 metasomatised Siberian SCLM is therefore enriched in Cl, Br and I by factors of up
114 to 125, 675 and 100 times, respectively, relative to the depleted MORB mantle
115 (DMM)²⁰. Thus, the SCLM is a significantly larger and more heterogeneous halogen
116 reservoir than previously considered and may impart a significant influence on global
117 volatile cycles²¹.

118

119 **Release of lithospheric halogens during LIP emplacement**

120 The comparatively large quantity of halogens residing in the base of the Siberian
121 SCLM means that even small proportions released to the surface will have
122 significant consequences for the global halogen cycle. The eruption of halogens into
123 the stratosphere catalyses ozone-destroying reactions, raising surface levels of
124 biologically damaging UV radiation^{22,23}. Transit of the SFB plume through the SCLM
125 could potentially have liberated significant amounts of halogens and other volatiles to
126 the atmosphere, contributing to species decline and extinction during the end-
127 Permian crisis.

128

129 Udachnaya xenoliths formed at significantly deeper depths (>50km difference)¹⁷ in
130 the lithosphere compared to Obnazhennaya. The identification of Udachnaya-like
131 metasomatic signatures in Obnazhennaya indicates that volatiles residing in the
132 metasomatised basal SCLM were mobilised and ascended to shallower parts of the
133 SCLM. Obnazhennaya xenoliths have trace element signatures within the range of
134 previously reported values of the SFB¹⁷, strong P-PGE depletions uncharacteristic of

135 cratonic lithosphere and Os isotopic compositions consistent with a formation age
136 similar to the time of plume impingement¹³. These characteristics indicate that the
137 part of the lithosphere sampled by the Obnazhennaya kimberlite represents the melt
138 residue of the SFB-plume (Fig. 4d)¹³. The identification of metasomatised SCLM
139 signatures within the Obnazhennaya xenoliths therefore suggests that as the SFB
140 plume impacted the base of the lithosphere, the resulting melts incorporated volatiles
141 mobilised from the deeper metasomatised SCLM, before being erupted at the
142 surface or stalling in the lithosphere. The contribution of SCLM-derived volatiles to
143 the SFB plume can therefore be estimated using differences in halogen and noble
144 gas signatures between the Udachnaya (metasomatised SCLM) and Obnazhennaya
145 (SFB + metasomatised SCLM) xenoliths.

146

147 Assuming that the melt residues in the Obnazhennaya lithosphere had a starting
148 composition similar to the SFB plume (12.7R_A and mantle-like Br/Cl and I/Cl)^{24,25}
149 then the amount of assimilation from the SCLM (Udachnaya) can be estimated from
150 the extent of mixing between the two sources (Fig. 2, 3, Supplementary Information).
151 Comparing He, Br and I systematics between the SFB plume and the SCLM
152 component represented by Udachnaya, requires that up to 70% of volatiles in
153 Obnazhennaya be derived from the SCLM (Fig. 3a,b, Supplementary Fig. 2).
154 Furthermore, any potential overprinting related to crustal assimilation affecting the
155 halogen composition of the melt can be excluded as the rapid transport of xenoliths
156 to the surface via kimberlite volcanism limits any potential interaction with the
157 surrounding crust^{26,27}.

158

159 Taking the volume of Cl degassed as calculated from the SFB melt inclusions
160 (8.7×10^{15} kg), then the total fluxes of Br and I to the atmosphere are estimated to be
161 2.3×10^{13} kg and 9.6×10^{10} kg respectively. This calculation assumes the melt had
162 Br/Cl and I/Cl values similar to Obnazhennaya and considers that halogens are not
163 fractionated during degassing²⁸. Explosive eruptions inject reactive HCl and HBr
164 gases into the lower stratosphere (~12-25 km) and deplete ozone levels, whereas
165 effusive eruptions lead to soluble HCl being washed-out prior to reaching the
166 stratosphere²⁴. Considering only explosive events (20-30% of the SFB)²⁹, a ~75%
167 rate of stratospheric injection³⁰, and the amount of Cl measured within the SFB³¹,
168 then the mass of Cl released to the stratosphere over the main eruptive phase of the
169 SFB (two thirds of the total eruptive volume over 300ky)³² is the equivalent to 0.5-1.0
170 Pinatubo (1991-1992) eruptions, which caused a 15-20% reduction in global
171 ozone³³, every year for 300ky. Models of ozone depletion during the SFB eruption,
172 using estimated stratospheric HCl fluxes predict a 30-55% reduction in ozone over
173 the same eruptive timeframe⁴. These stratospheric HCl flux estimates are 5 times
174 lower than the predicted from the SFB melt inclusions (8.7×10^{15} kg)³¹. Furthermore,
175 these estimates do not take into account the consequences of Br degassing on
176 ozone depletion. The large release of Br to the stratosphere during the SFB, as
177 indicated by the high Br/Cl of the Siberian SCLM, likely further exacerbated ozone
178 depletion. Bromine has a much greater capacity for depleting ozone (~45× more
179 effective²⁴) and could have reduced ozone levels by a further 20% during SFB
180 eruption. Although there are several uncertainties in the rate and magnitude of
181 volatile degassing during SFB magmatism, the scale of halogen degassing fluxes
182 presented here are sufficient to incur a near to total loss of global ozone during the
183 end-Permian crisis.

184 Melt inclusions within the SFB contain between 0.01-0.33 wt.% Cl^{10,31}, which is an
185 order of magnitude higher than the maximum measured Cl concentrations in other
186 LIPs^{8,31}. Inclusions with high Cl concentrations are found to be equally enriched in
187 other volatiles species such as fluorine (1.95 wt. %) and sulphur (0.51 wt. %)³¹. To
188 create such high volatile contents within these melts, starting from a DMM-like
189 composition, would require very low degree partial melting, or the assimilation of
190 volatiles from another unknown reservoir. Low degrees partial melting can
191 concentrate volatiles in the melt fraction, however the high Mg contents measured
192 within the melt inclusions preclude low degree partial melting, suggesting that the
193 assimilation of volatile-rich material is most likely cause of the high volatile contents
194 in the SFB.

195

196 The composition of the Obnazhennaya xenoliths, assumed to represent plume melt
197 residues, can be used to estimate the pre-eruptive SFB melt composition and
198 establish if the SCLM is the potential source of the SFB volatile enrichment.
199 Obnazhennaya xenoliths require ~30% melt extraction to account for the elevated
200 $F_{O>92}$ of the olivines¹³ and therefore the Cl composition of this melt can be estimated
201 using a batch melting model³⁴. Experimentally determined partition coefficients for Cl
202 between olivine ($D_{Cl}^{Ol/Melt} = 1.9 \times 10^{-2}$) and pyroxene ($D_{Cl}^{Pyx/Melt} = 1.5 \times 10^{-2}$) were used
203 to calculate the Cl concentration of the melt at 1500°C, prior to eruption³⁵. Using the
204 range of Cl concentrations in the olivine and pyroxene minerals from Obnazhennaya
205 xenoliths yield Cl concentrations of 0.1-0.2 wt.% in the melt. These estimates are
206 considered upper limits given the potential for an unknown proportion of intact fluid
207 inclusions to remain following crushing, thus Cl data based on stepped heating is
208 likely to overestimate the Cl abundance within the minerals. However, it is notable

209 that the melt Cl estimates are consistent with previous values for the eruptive melt
210 composition (0.01-0.33 wt.% Cl)^{10,32} providing confidence in the assumptions we
211 have made and confirming that the SFB melt was already enriched in Cl prior to
212 eruption. These arguments therefore provide further (albeit indirect) support for a
213 SCLM origin for the majority of halogens in SFB melts.

214

215 **Implications for the end-Permian extinction**

216 Ozone depletion during the end-Permian crisis is considered to have led to the
217 decline in the dominant terrestrial plant species at the time, followed by the rapid
218 expansion of opportunistic lycopsids³⁶. The global distribution of preserved
219 microspores from these emerging lycopsides exhibit features indicative of a failure in
220 the normal development process of the spores. The global dispersion of these
221 mutagenic spores suggest that this was a reaction to global stress factors, unlikely to
222 be related to changes in global temperature from the release of gases such as SO₂
223 and CO₂ during SFB emplacement⁴. Experiments on the effects of end-Permian UV-
224 B regimes on modern conifers led to a fivefold increases in the occurrence of
225 mutagenic malformations, and complete sterilisation³⁷. This would have caused
226 widespread deforestation and the collapse of the terrestrial biosphere, indicating
227 ozone depletion was a major contributing factor in the end-Permian mass extinction
228 event^{4,37}.

229

230 The peak occurrence of mutagenic spores occurs prior to the rapid negative shift in
231 $\delta^{13}\text{C}$ in end-Permian carbonates attributed to the extinction of calcified marine life³⁶.

232 The $\delta^{13}\text{C}$ excursions coincide with a change from predominantly extrusive to
233 intrusive eruptions of the Siberian LIP¹. The emplacement of sills into volatile rich
234 sediments was considered to have released vast quantities of volatiles including CO₂
235 and halocarbons gases to the atmosphere leading to rapid climate change and
236 ozone depletion^{1,5}. However, from the palynological evidence³⁶ it is clear that a
237 reduction in terrestrial biodiversity was occurring prior to the onset of marine
238 extinction. Furthermore, evidence for reduced sedimentation rates prior to and up to
239 the PTB, indicates that there was global eustatic sea-level regression potentially
240 caused by falling global temperatures and the onset of glaciation³⁸. The rapid fall in
241 temperatures has been linked to the emission of SO₂ to the atmosphere during the
242 eruptive phase of the SFB⁷.

243

244 The concurrent timing of the eruptive phase of the SFB and the palynological
245 evidence for ozone depletion is not consistent with the idea of sedimentary brines
246 degassing during later intrusive phases of igneous activity being the primary source
247 of halogens causing ozone destruction. As we have shown in this study the majority
248 of halogens in the SFB were added during plume-lithosphere interaction followed by
249 their subsequent release to the atmosphere during explosive eruptions. Sulphur
250 enrichments, co-existing with halogens in SFB³¹, may also have been derived from
251 the SCLM (Fig. 4c). Evidence for terrestrial species decline prior to PTB therefore
252 suggests that the release of halogens and gaseous sulphur species, and the
253 subsequent decline in ozone and global temperatures respectively were the
254 predominant factors in initiating the end-Permian mass extinction. The change in
255 eruptive phase from explosive to intrusive may have played a role in extending the
256 extinction from a mainly terrestrial phenomenon to a global extinction event.

257

258 **Subducted origin of volatiles in the Siberian lithosphere**

259 The high concentrations of halogens in the Udachnaya xenoliths indicate the
260 Siberian SCLM has been enriched in volatiles by metasomatic processes.
261 Udachnaya xenoliths have Br/Cl and I/Cl similar to fluids trapped within minerals in
262 the altered oceanic crust (AOC), suggesting that the metasomatism of the Siberian
263 SCLM was potentially driven by subduction derived fluids (Fig. 2)³⁹. Combined with
264 the noble gases (Fig. 3; Supplementary Fig. 3 and 4) the Udachnaya xenoliths show
265 an evolution from seawater-like $^3\text{He}/^4\text{He}$, Br/Cl and I/Cl values, to values with
266 increasingly radiogenic $^3\text{He}/^4\text{He}$ and enriched Br/Cl and I/Cl (Fig. 3), further
267 suggesting that the metasomatic fluid originated as seawater but subsequently
268 evolved, during subduction or within the SCLM, due to halogen fractionation and the
269 production of ^4He from U-Th decay¹⁸. Eclogite xenoliths from the Udachnaya
270 kimberlite exhibit $\delta^{18}\text{O}$ up to +7.7%⁴⁰, outwith the normal mantle range (+5.4 \pm
271 0.2%)⁴¹ indicating that they originated as oceanic crust that underwent low
272 temperature alteration⁴⁰. Eclogites formed from the subduction of oceanic crust have
273 been shown to retain the halogen and oxygen isotopic signature of the oceanic crust
274 protolith during metamorphism, providing a mechanism for the delivery of
275 subduction-derived halogens to the Siberian SCLM^{42,43}.

276

277 As we have shown in this study, up to 70% of the volatile content of the Siberian
278 plume originated from assimilation of metasomatised lithospheric material. The
279 composition of the SCLM therefore plays an integral role in controlling the volatile
280 content of LIPs, and as such the overall effect they have on the global environment.

281 Based on the evolution of the Br/Cl, I/Cl and $^3\text{He}/^4\text{He}$ of the xenoliths from seawater
282 to AOC-like values, coupled with the AOC-like $\delta^{18}\text{O}$ within eclogite xenoliths from
283 Udachnaya, it appears that the origin of the Siberian SCLM volatiles is from the
284 subduction of a seawater derived component within the AOC. Enrichment of
285 seawater-derived volatiles in the Siberian SCLM provided the plume with an
286 abundant supply of halogens, which were released to the atmosphere during
287 eruption and resulted in globally extensive reductions in ozone levels and the decline
288 of the biosphere. The SCLM is also a major repository for other subducted volatile
289 species including sulphur and carbon^{44,45}, which can also contribute to environmental
290 degradation during plume-lithosphere interaction and LIP emplacement⁷. The SCLM
291 can therefore act as a repository of subducted volatile that can periodically be
292 mobilised and released to the Earth's surface and atmosphere during deep-seated
293 melting and volcanism leading to devastating impacts on the global environment.

294

295 **References**

- 296 1 Burgess, S., Muirhead, J. & Bowring, S. Initial pulse of Siberian Traps sills as the
297 trigger of the end-Permian mass extinction. *Nature Communications* **8** (2017).
- 298
- 299 2 Wignall, P. B. Large igneous provinces and mass extinctions. *Earth-science reviews*
300 **53**, 1-33 (2001).
- 301
- 302 3 Erwin, D. H., Bowring, S. A. & Yugan, J. End-Permian mass extinctions: a review.
303 *Special Paper - Geological Society of America*, 363-384 (2002).

304

- 305 4 Beerling, D. J., Harfoot, M., Lomax, B. & Pyle, J. A. The stability of the stratospheric
306 ozone layer during the end-Permian eruption of the Siberian Traps. *Philosophical*
307 *Transactions of the Royal Society of London A: Mathematical, Physical and*
308 *Engineering Sciences* **365**, 1843-1866 (2007).
309
- 310 5 Svensen, H. *et al.* Siberian gas venting and the end-Permian environmental crisis.
311 *Earth and Planetary Science Letters* **277**, 490-500 (2009).
312
- 313 6 Grard, A., Francois, L., Dessert, C., Dupré, B. & Godderis, Y. Basaltic volcanism and
314 mass extinction at the Permo-Triassic boundary: environmental impact and modeling
315 of the global carbon cycle. *Earth and Planetary Science Letters* **234**, 207-221 (2005).
316
- 317 7 Guex, J. *et al.* Thermal erosion of cratonic lithosphere as a potential trigger for mass-
318 extinction. *Scientific reports* **6**, 23168 (2016).
319
- 320 8 Self, S., Widdowson, M., Thordarson, T. & Jay, A. E. Volatile fluxes during flood
321 basalt eruptions and potential effects on the global environment: A Deccan
322 perspective. *Earth and Planetary Science Letters* **248**, 518-532 (2006).
323
- 324 9 Sobolev, A., Sobolev, S., Kuzmin, D., Malitch, K. & Petrunin, A. Siberian
325 meimechites: origin and relation to flood basalts and kimberlites. *Russian Geology*
326 *and Geophysics* **50**, 999-1033 (2009).
327
- 328 10 Sobolev, S. V. *et al.* Linking mantle plumes, large igneous provinces and
329 environmental catastrophes. *Nature* **477**, 312 (2011).

330

331 11 Reichow, M. K. *et al.* $^{40}\text{Ar}/^{39}\text{Ar}$ dates from the West Siberian Basin: Siberian flood
332 basalt province doubled. *Science* **296**, 1846-1849 (2002).

333

334 12 Ivanov, A. V. *et al.* Siberian Traps large igneous province: Evidence for two flood
335 basalt pulses around the Permo-Triassic boundary and in the Middle Triassic, and
336 contemporaneous granitic magmatism. *Earth-Science Reviews* **122**, 58-76 (2013).

337

338 13 Pernet-Fisher, J. *et al.* Plume impingement on the Siberian SCLM: Evidence from
339 Re–Os isotope systematics. *Lithos* **218**, 141-154 (2015).

340

341 14 Walker, R., Carlson, R., Shirey, S. & Boyd, F. Os, Sr, Nd, and Pb isotope
342 systematics of southern African peridotite xenoliths: implications for the chemical
343 evolution of subcontinental mantle. *Geochimica et Cosmochimica Acta* **53**, 1583-
344 1595 (1989).

345

346 15 McDonough, W. Constraints on the composition of the continental lithospheric
347 mantle. *Earth and Planetary Science Letters* **101**, 1-18 (1990).

348

349 16 Taylor, L. A., Milledge, H. J., Bulanova, G. P., Snyder, G. A. & Keller, R. A.
350 Metasomatic eclogitic diamond growth: evidence from multiple diamond inclusions.
351 *International Geology Review* **40**, 663-676 (1998).

352

353 17 Howarth, G. H. *et al.* Superplume metasomatism: evidence from Siberian mantle
354 xenoliths. *Lithos* **184**, 209-224 (2014).

355

356 18 Barry, P. H. *et al.* Helium isotopic evidence for modification of the cratonic
357 lithosphere during the Permo-Triassic Siberian flood basalt event. *Lithos* **216**, 73-80
358 (2015).

359

360 19 Griffin, W., Fisher, N., Friedman, J., O'Reilly, S. Y., & Ryan, C. (2002). Cr-
361 pyrope garnets in the lithospheric mantle 2. Compositional populations and
362 their distribution in time and space. *Geochemistry, Geophysics, Geosystems*,
363 3(12), 1-35.

364

365 20 Kendrick, M. *et al.* Seawater cycled throughout Earth's mantle in partially
366 serpentinized lithosphere. *Nature Geoscience* **10**, 222-228 (2017).

367

368 21 Burgess, R., Layzelle, E., Turner, G. & Harris, J. Constraints on the age and halogen
369 composition of mantle fluids in Siberian coated diamonds. *Earth and Planetary*
370 *Science Letters* **197**, 193-203 (2002).

371

372 22 Johnston, D. A. Volcanic contribution of chlorine to the stratosphere: more significant
373 to ozone than previously estimated? *Science* **209**, 491-493 (1980).

374

375 23 Daniel, J., Solomon, S., Portmann, R. & Garcia, R. Stratospheric ozone destruction:
376 The importance of bromine relative to chlorine. *Journal of Geophysical Research:*
377 *Atmospheres* **104**, 23871-23880 (1999).

378

379 24 Kendrick, M. A., Kamenetsky, V. S., Phillips, D. & Honda, M. Halogen systematics

- 380 (Cl, Br, I) in mid-ocean ridge basalts: a Macquarie Island case study. *Geochimica et*
381 *Cosmochimica Acta* **81**, 82-93 (2012).
- 382
- 383 25 Basu, Asish R., et al. High-³He plume origin and temporal-spatial evolution of the
384 Siberian flood basalts. *Science* **269** (5225), 822-825 (1995).
- 385
- 386 26 Kelley, S. & Wartho, J. Rapid kimberlite ascent and the significance of Ar-Ar ages in
387 xenolith phlogopites. *Science* **289**, 609-611 (2000).
- 388
- 389 27 Alexeev, S. *et al.* Isotopic composition (H, O, Cl, Sr) of ground brines of the Siberian
390 Platform. *Russian Geology and Geophysics* **48**, 225-236 (2007).
- 391
- 392 28 Black, B. A., Elkins-Tanton, L. T., Rowe, M. C. & Peate, I. U. Magnitude and
393 consequences of volatile release from the Siberian Traps. *Earth and Planetary*
394 *Science Letters* **317**, 363-373 (2012).
- 395
- 396 29 Aiuppa, A. *et al.* Emission of bromine and iodine from Mount Etna volcano.
397 *Geochemistry, Geophysics, Geosystems* **6** (2005).
- 398
- 399 30 Ross, P.-S. *et al.* Mafic volcanoclastic deposits in flood basalt provinces: a review.
400 *Journal of Volcanology and Geothermal Research* **145**, 281-314 (2005).
- 401
- 402 31 Millard, G. A., T. A. Mather, D. M. Pyle, William I. Rose, and B. Thornton. "Halogen
403 emissions from a small volcanic eruption: Modeling the peak concentrations,
404 dispersion, and volcanically induced ozone loss in the stratosphere." *Geophysical*

405 *Research Letters* **33(19)** (2006).

406

407 32 Burgess, S. D., & Bowring, S. A. High-precision geochronology confirms voluminous
408 magmatism before, during, and after Earth's most severe extinction. *Science*
409 *Advances*, 1(7) (2015).

410

411 33 Westrich, H. R., & Gerlach, T. M. Magmatic gas source for the stratospheric SO₂
412 cloud from the June 15, 1991, eruption of Mount Pinatubo. *Geology*, **20**(10), 867-870
413 (1992).

414

415 34 Shaw, D. M. Trace elements in magmas: a theoretical treatment, Cambridge
416 University Press (2006).

417

418 35 Joachim, B. *et al.* Experimental partitioning of F and Cl between olivine,
419 orthopyroxene and silicate melt at Earth's mantle conditions. *Chemical Geology* **416**,
420 65-78 (2015).

421

422 36 Visscher, H., Looy, C. V., Collinson, M. E., Brinkhuis, H., Van Konijnenburg-Van
423 Cittert, J. H., Kürschner, W. M., & Sephton, M. A. Environmental mutagenesis during
424 the end-Permian ecological crisis. *Proceedings of the National Academy of Sciences*
425 *of the United States of America*, 101(35), 12952-12956 (2004).

426

427 37 Benca, Jeffrey P., Ivo AP Duijnste, and Cindy V. Looy. "UV-B-induced forest
428 sterility: Implications of ozone shield failure in Earth's largest extinction." *Science*
429 *advances* **4.2** (2018)

430

431 38 Baresel, B., Bucher, H., Bagherpour, B., Brosse, M., Guodun, K., & Schaltegger, U.
432 Timing of global regression and microbial bloom linked with the Permian-Triassic
433 boundary mass extinction: implications for driving mechanisms. *Scientific Reports*, **7**,
434 43630 (2017).

435

436 39 Chavrit, D. *et al.* The contribution of hydrothermally altered ocean crust to the mantle
437 halogen and noble gas cycles. *Geochimica et Cosmochimica Acta* **183**, 106-124
438 (2016).

439

440 40 Jacob, D., Jagoutz, E., Lowry, D., Matthey, D. & Kudrjavitseva, G. Diamondiferous
441 eclogites from Siberia: remnants of Archean oceanic crust. *Geochimica et*
442 *Cosmochimica Acta* **58**, 5191-5207 (1994).

443

444 41 Eiler, J. M. Oxygen isotope variations of basaltic lavas and upper mantle rocks.
445 *Reviews in mineralogy and geochemistry* **43**, 319-364 (2001).

446

447

448 42 Svensen, H., Jamtveit, B., Banks, D. A. & Austrheim, H. Halogen contents of eclogite
449 facies fluid inclusions and minerals: Caledonides, western Norway. *Journal of*
450 *Metamorphic Geology* **19**, 165-178 (2001).

451

452 43 Philippot, P., Agrinier, P., & Scambelluri, M. Chlorine cycling during subduction of
453 altered oceanic crust. *Earth and Planetary Science Letters*, **161**(1-4), 33-44 (1998).

454

455 44 Callegaro, S. *et al.* Microanalyses link sulfur from large igneous provinces
456 and Mesozoic mass extinctions. *Geology* **42**, 895–898 (2014).

457

458 45 Foley, S.F. and Fischer, T.P., An essential role for continental rifts and
459 lithosphere in the deep carbon cycle. *Nature Geoscience*, **10**(12), 897-902
460 (2017)

461

462 46 Muramatsu, Y. *et al.* Halogen concentrations in pore waters and sediments of the
463 Nankai Trough, Japan: Implications for the origin of gas hydrates. *Applied*
464 *Geochemistry* **22**, 534-556 (2007).

465

466 **Acknowledgments**

467 This work is dedicated to L. A. T, who passed away in 2017. L. A. T. devoted his life
468 to science and teaching, serving as an excellent mentor to P. H. B. during his time at
469 University of Tennessee. This work was financially supported through a NERC
470 studentship NE/J500057/1 (to M. W. B.) and a NERC (NE/M000427/1) and ERC
471 (ERC-267692 NOBLE) grant to C. J. B. and R. B. P. H. B. was funded by an NSF
472 fellowship (EAR-114455) to investigate the geochemical signatures in these
473 samples.

474

475 **Author contributions**

476 M. W. B, P. H. B and R. B conceived the project and prepared the initial manuscript.
477 L.A.T provided the samples and M. W. B and R. B performed the analysis. All
478 authors contributed to data analysis and preparation of the final manuscript.

479

480 **Competing financial interests**

481 The authors declare no competing financial interests.

482

483 **Figure Captions**

484 **Figure 1. Halogen and K abundances in Udachnaya and Obnazhennaya**
485 **xenoliths.** Abundances of (a) Br, (b) I and (c) K plotted against Cl showing the
486 enrichment of Br and I within the Siberian SCLM relative to the MORB/OIB mantle
487 source and seawater²⁴. Crushing (“crush”) data are shown by open symbols and
488 stepped-heating (“melt”) by closed symbols. Errors bars are consistently covered by
489 symbol. Uncertainties are presented at 1σ .

490

491 **Figure 2. Halogen composition of the Siberian SCLM.** Br/Cl versus I/Cl (a)
492 crushing and (b) step heating of the Udachnaya and Obnazhennaya xenoliths.
493 Xenoliths show a range of halogen compositions which overlap the range of Br/Cl
494 and I/Cl observed in altered oceanic crust (AOC fluids) and eclogites. Seawater
495 evaporation trend (SET) is shown, indicating that sedimentary brines cannot be
496 responsible for the halogen signature of the xenoliths. Br/Cl and I/Cl results from
497 halogen fractionation (black arrows) of AOC fluids (green star). Seawater, marine

498 pore fluids and MORB/OIB compositions are shown for reference^{24,46}. Figure
499 symbols are the same as in Figure 1. Uncertainties are 1σ .

500

501 **Figure 3. Helium isotopes and halogen systematics.** $^3\text{He}/^4\text{He}$ versus **(a)** I/Cl and
502 **(b)** Br/Cl from crushed release of Udachnaya and Obnazhennaya xenoliths.

503 Udachnaya has $^3\text{He}/^4\text{He}$, Br/Cl and I/Cl, which range from values similar to seawater
504 (blue star) towards higher Br/Cl and I/Cl and lower $^3\text{He}/^4\text{He}$ characteristic subduction
505 modified seawater. Obnazhennaya has higher $^3\text{He}/^4\text{He}$ ranging from Udachnaya-like
506 towards OIB-like values, from the influx of plumes melts. Mixing lines [$r =$
507 $(^4\text{He}/\text{Cl})_{\text{plume}} / (^4\text{He}/\text{Cl})_{\text{SCLM}}$] shown between a hypothetical SCLM component (black
508 diamond, intercept through the data) and plume melts with the relative percentage of
509 SCLM assimilation shown. Figure symbols are same as used in Figure 1.

510 Uncertainties are 1σ .

511

512 **Figure 4. Schematic of plume-lithosphere interaction within the Siberian**
513 **craton.** a) SCLM is composed partly of metasomatised peridotite from addition of
514 subducted volatiles, which potentially seeds diamond formation. b) Intermittent
515 influence of the Siberian plume drives kimberlite volcanism c) Plume melt impinges
516 on the lithosphere, incorporating volatile rich SCLM material. Halogens released to
517 the atmosphere during explosive SFB eruptions leading to the extensive ozone
518 depletion. d) Plume retreats leaving a much-reduced SCLM with veins of melt
519 residue, followed by a second period of kimberlitic volcanism, transporting melt
520 residues and SCLM material to the surface.

521

522 **Methods**

523 **Neutron irradiation noble gas mass spectrometry (NI-NGMS)**

524 Olivine and clinopyroxene mineral separates from Udachnaya and Obnazhennaya
525 peridotite xenoliths were selected for heavy halogen (Cl, Br and I) and K analysis
526 using neutron-irradiated noble gas mass spectrometry (NI-NGMS)⁴⁷⁻⁵². Samples
527 weighing between 0.015 and 0.062 g were first cleaned with deionised water in an
528 ultrasonic bath for 20 minutes, followed by a further 5 minutes in acetone. Samples
529 were then dried under a heat lamp at 100°C, wrapped in Al foil and sealed in
530 evacuated fused-silica tubes together with Hb3gr, scapolite and Shallowater
531 meteorite standards to monitor noble gas proxy production from K and halogens⁵¹.

532

533 Samples were irradiated in the GRICIT (MN2014b) facilities of the TRIGA Reactor,
534 Oregon State University for a few hours each day between 22/04/2014 and
535 01/07/2014 giving a total irradiation time of 205 hours. Irradiation details and monitor
536 values for this irradiation have been reported previously⁵¹.

537

538 Noble gas proxy isotopes (³⁸Ar_{Cl}, ⁸⁰Kr_{Br}, ¹²⁸Xe_I and ³⁹Ar_K) formed during irradiation as
539 well as natural Ar, Kr and Xe isotopes were measured on the MS1 mass
540 spectrometer⁵². A subset of samples and repeat analyses were also performed on a
541 Thermo Fisher Scientific ARGUS VI mass spectrometer⁵¹. Noble gases were firstly
542 extracted from trapped fluid inclusions by loading samples into hand-operated

543 modified Nupro[®] valve crushers⁵³ (MS1 mass spectrometer). For bulk sample
544 analysis, powders from crushing analyses were loaded into in a tantalum resistance
545 furnace (MS1) and step heated using four temperature steps of 600°C, 1000°C,
546 1400°C and 1600°C to release halogens contained within the mineral matrix.
547 Halogens from four samples (UV33, UV88, UV357 and O129-74) were extracted
548 using a 10.6 µm wavelength CO₂ laser (Teledyne CETAC Fusions CO₂ – ARGUS
549 VI). In order to test that both extraction methods gave the similar results, sample
550 O97-12 was analysed using both the furnace and laser, Br/Cl and I/Cl varied by less
551 than 20% and 35% respectively between laser and furnace extraction suggesting
552 halogens were released in similar proportions using both extraction methods.
553 Halogens abundances were then calculated using the well-defined conversion
554 standards with known halogen concentrations (Hb3gr, scapolite and Shallowater
555 meteorite)^{50,51}, which monitor the efficiency of noble gas production through thermal
556 and epithermal neutron reactions.

557

558 Air calibrations and blanks were analysed daily to check the sensitivity and
559 background of the spectrometers, with maximum furnace blank values at 1600°C on
560 the MS1 being $1.65 \times 10^{-10} \text{ cm}^3 \text{ STP } ^{40}\text{Ar}$, $2.92 \times 10^{-13} \text{ cm}^3 \text{ STP } ^{84}\text{Kr}$ and 3.54×10^{-14}
561 $\text{STP cm}^3 \text{ } ^{132}\text{Xe}$ and ARGUS VI blanks being $5.76 \times 10^{-12} \text{ cm}^3 \text{ STP } ^{40}\text{Ar}$ and $1.41 \times$
562 $10^{-15} \text{ cm}^3 \text{ STP } ^{132}\text{Xe}$, with Kr blanks below detection limit. Noble gas purification
563 analytical procedures for the MS1 and ARGUS VI mass spectrometers as well as the
564 data reduction procedures have been documented previously^{52,56}. External precision
565 is reported at 3% (1σ) for Cl and 7% (1σ) for Br and I determinations.

566

567 **Data availability**

568 All data pertaining to this study is presented in the paper and the supplementary
569 information. Correspondence and requests for materials should be addressed to the
570 corresponding author.

571

572 **Method References**

573 47 Merrihue, C. & Turner, G. Potassium-argon dating by activation with fast neutrons.
574 *Journal of Geophysical Research* 71, 2852-2857 (1966).

575

576 48 Böhlke, J. & Irwin, J. Laser microprobe analyses of Cl, Br, I, and K in fluid inclusions:
577 Implications for sources of salinity in some ancient hydrothermal fluids. *Geochimica*
578 *et Cosmochimica Acta* 56, 203-225 (1992).

579

580 49 Johnson, L., Burgess, R., Turner, G., Milledge, H. & Harris, J. Noble gas and
581 halogen geochemistry of mantle fluids: comparison of African and Canadian
582 diamonds. *Geochimica et Cosmochimica Acta* 64, 717-732 (2000).

583

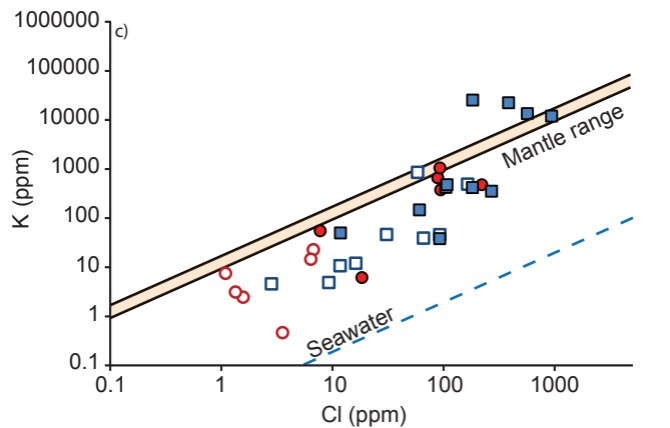
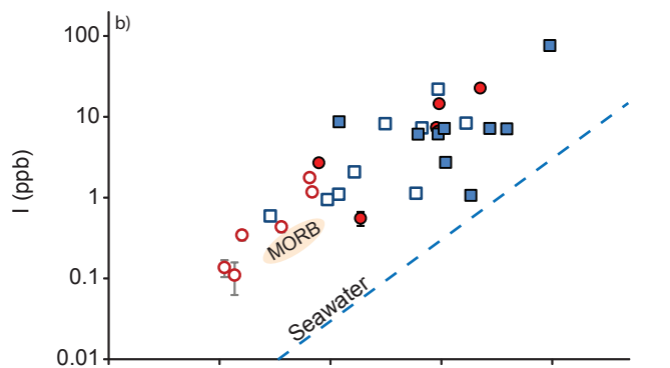
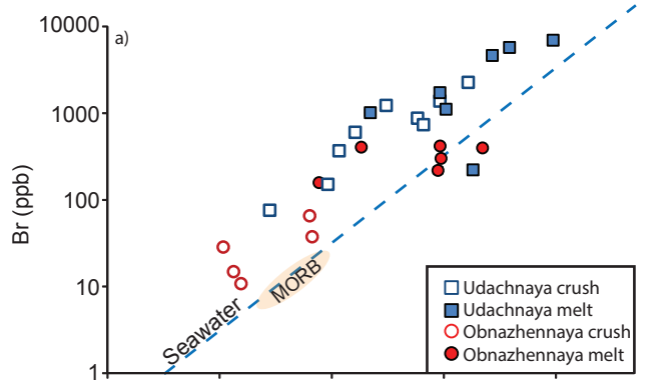
584 50 Kendrick, M. A. High precision Cl, Br and I determinations in mineral standards using
585 the noble gas method. *Chemical Geology* 292, 116-126 (2012).

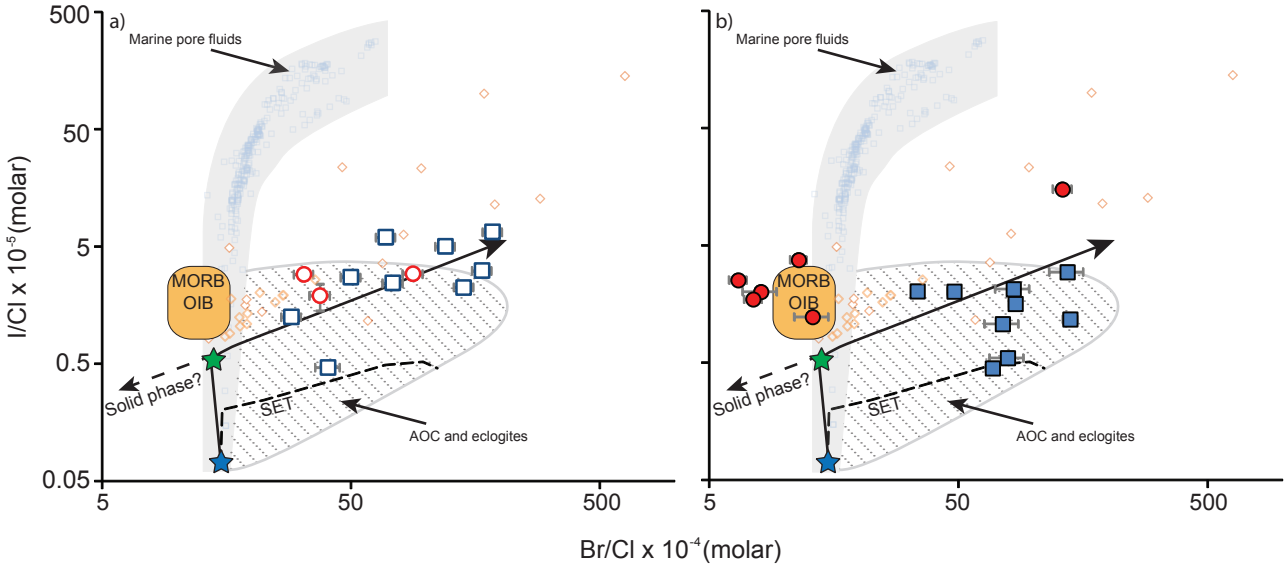
586

587 51 Ruzié-Hamilton, L. et al. Determination of halogen abundances in terrestrial and
588 extraterrestrial samples by the analysis of noble gases produced by neutron
589 irradiation. *Chemical Geology* 437, 77-87 (2016).

590

- 591 52 Broadley, M. W., Ballentine, C. J., Chavrit, D., Dallai, L. & Burgess, R. Sedimentary
592 halogens and noble gases within Western Antarctic xenoliths: Implications of
593 extensive volatile recycling to the sub continental lithospheric mantle. *Geochimica et*
594 *Cosmochimica Acta* 176, 139-156 (2016).
595
- 596 53 Stuart, F. & Turner, G. The abundance and isotopic composition of the noble gases
597 in ancient fluids. *Chemical Geology: Isotope Geoscience section* 101, 97-109 (1992).
598

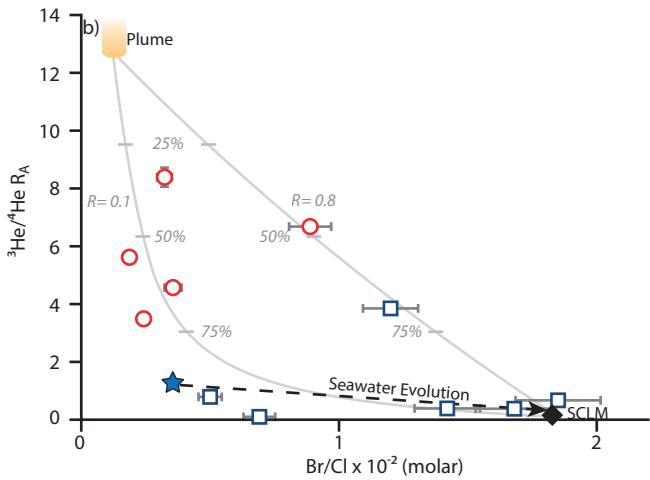
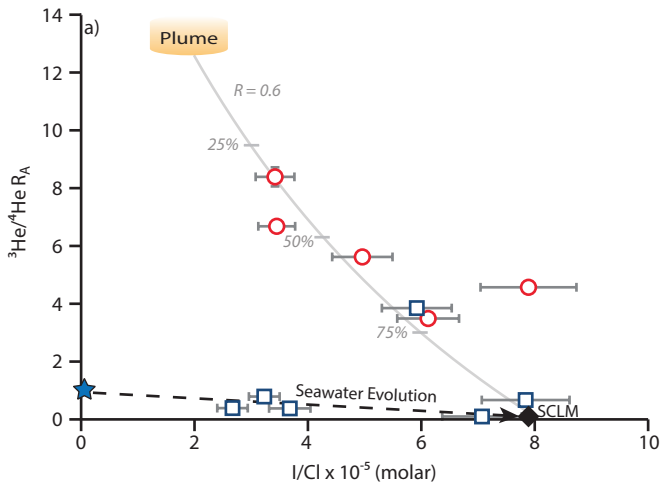




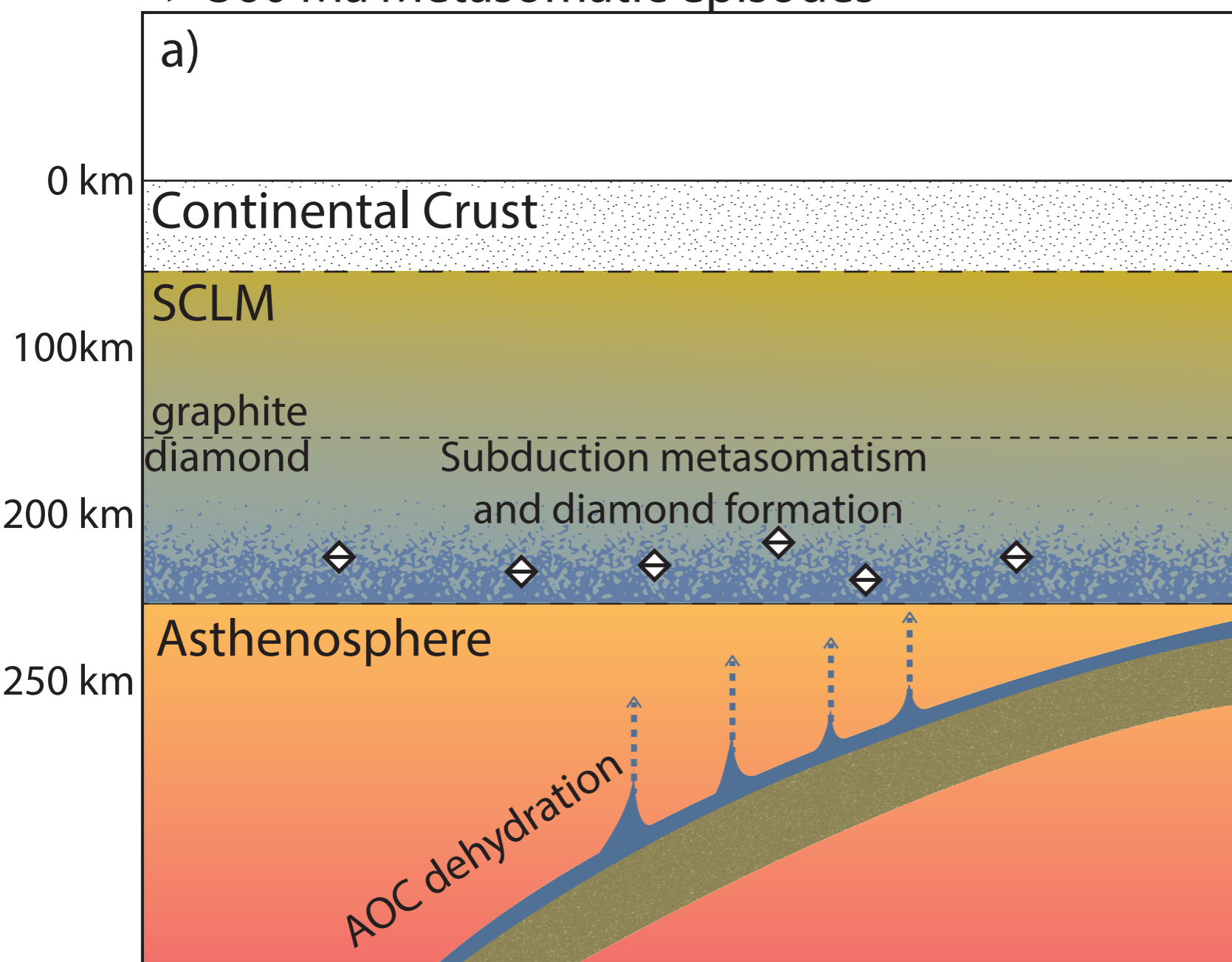
★ Starting fluid

★ Modern seawater

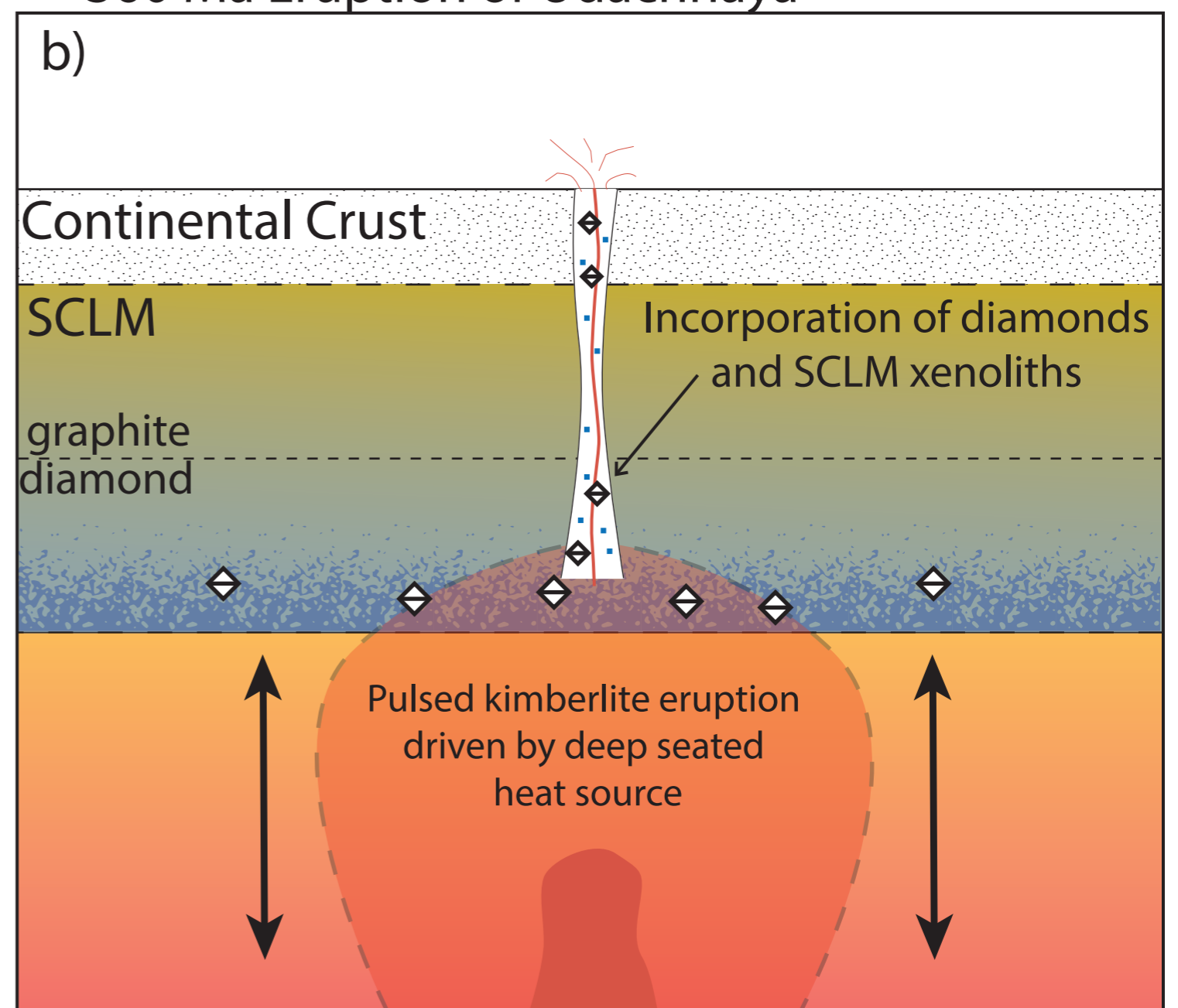
◇ Canadian diamonds



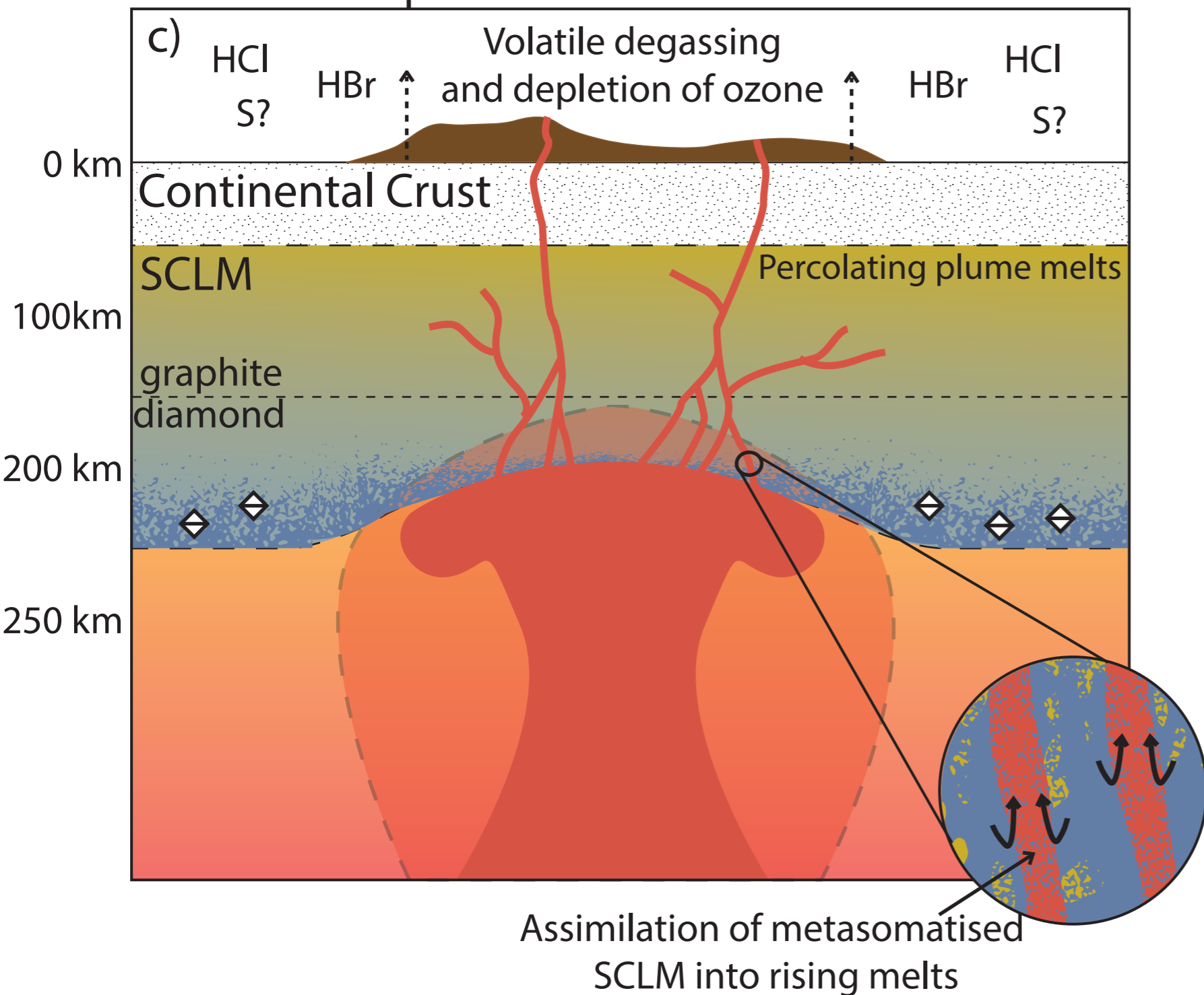
> 360 Ma Metasomatic episodes



~ 360 Ma Eruption of Udachnaya



~ 250 Ma Eruption of Siberian Flood Basalts



~ 160 Ma Eruption of Obnazhennaya

